## <u>Sub-atomic Particles – Revision Worksheet</u>

1. Give one word for the following:						
1.1 Particles with a positive charge.						
1.2 Particles that are found around the nucleus.						
1.3 Particles without a charge that are found in the nucleus.						
1.4 Particles responsible for the mass of an atom.						
1.5 Particles that determine the volume of an atom.						
2. Draw a labelled diagram of a Boron atom.						

## 3. Complete the table to indicate all the particles of an atom.

Particle	Charge	Where does it occur?	
	Positive		
Neutron			
		Around the nucleus	

4. Give the follow	ving definitio	ns:			
4.1 Proton					
4.2 Atom					
4.3 Electron					
4.4 Neutron					
5. Look at the fol	lowing sketcl	h of a neutral a	tom and answe	er the questi	ons:
5.1 What is the nucl	eus made up of	?	<u> </u>	$\overline{}$	
5.2 Which particles	move in the spa	ce around the nuc	cleus?	9	(5)X
5.3 How many of the	e following part	icles are in this ato	om: <b>(</b>	+	$\sim$
5.3.1 neutrons?				/	
<ul><li>5.3.2 electrons?</li><li>5.3.3 protons?</li></ul>					_
2.2.2 protons:					